PHYS 206:

KOÇ UNIVERSITY

Spring Semester 2013

College of Arts and Sciences

Sectiong Quiz 7

29 March 2013

Closed book. No calculators are to be used for this quiz. Quiz duration: 10 minutes

Name:	Student ID:	Signature:
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For a particular cathode material in a photoelectric-effect experiment, you measure stopping potentials V₀=1.0 V for light of wavelength λ =600 nm, 2.0 V for 400 nm, and 3.0 V for 300 nm. Determine the work function φ for this material and the implied value of Planck's constant *h*.

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Name:	Student ID:	Signature:
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The photoelectric threshold wavelength of a tungsten surface is 272 nm. Calculate the maximum kinetic energy of the electrons ejected from this tungsten surface by ultraviolet radiation of frequency 1.45×10^{15} Hz Express the answer in electron volts.

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Name:	Student ID:	Signature:
Protons are accelerated	d from rest by a potential of	difference of 4.00 kV and strike a metal
target. If a proton pro	duces one photon on impa	ict, what is the minimum wavelength of

the resulting X-rays? How does your answer compare to the minimum wavelength if 4.00keV electrons are used instead? Why do x-ray tubes use electrons rather than protons to produce x rays?

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If a photon of wavelength 0.04250 nm strikes a free electron and is scattered at an angle of 35.0° from its original direction, find a) the change in the wavelength of this photon; b) the wavelength of the scattered light: c) the change in energy of the photon (is it a loss or a gain?); d) the energy gained by the electron.

$$\lambda' - \lambda = \frac{h}{mc} (1 - \cos\varphi)$$

PHYS206: General Physics IV **KOÇ UNIVERSITY**

Spring Semester 2013

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Many varieties of lasers emit light in the form of pulses rather than a steady beam. A
tellurium-sapphire laser can produce light at wavelength of 800 nm in ultrashort pulses
that last only 4.00×10^{-15} s (4.00 femtoseconds, or 4.00 fs). The energy in a single pulse
produced by one such laser is $2.00 \ \mu J = 2.00 \times 10^{-6} J$, and the pulses propagate in the
positive x-direction. Find a) the frequency of the light; b) the energy and minimum
energy uncertainty of a single photon in the pulse; c) the minimum frequency uncertainty
of the light in the pulse; d) the spatial length of the pulse, in meters and as a multiple of
the wavelength; e) the momentum and minimum momentum uncertainty of a single

photon in the pulse and f) the approximate number of photons in the pulse.

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The photoelectric work functions for particular samples of certain metals are as follows: cesium, 2.1 eV; copper, 4.7 eV; potassium, 2.3 eV; and zinc, 4.3 eV. a) What is the threshold wavelength for each metal surface? b) Which of these metals could not emit photoelectrons when irradiated with visible light (380-750 nm)?